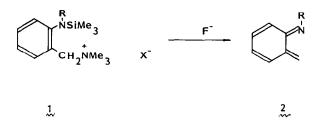
A NEW APPROACH TO GEPHYROTOXIN

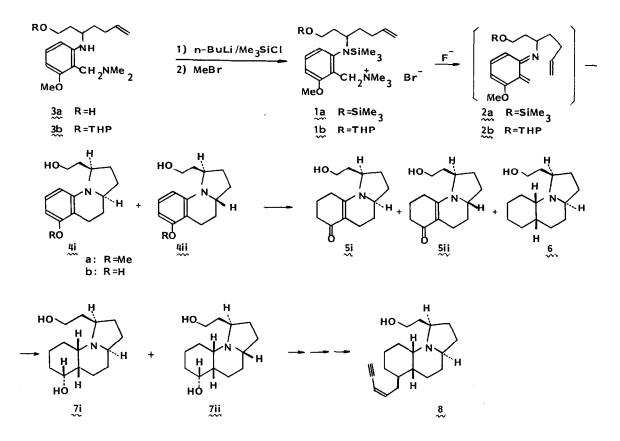
Y. Ito, E. Nakajo, M. Nakatsuka and T. Saegusa Department of Synthetic Chemistry, Faculty of Engineering Kyoto University, Kyoto, Japan 606

<u>Summary</u>: CsF induced 1,4-elimination of ammonium bromide <u>la</u> and the subsequent intramolecular cyclization of the resulting o-quinone methide N-alkenylimine <u>2a</u> afforded benzo[e]indolizidine <u>4a</u> as a 4~5 : 1 stereoisomeric mixture. Hydrogenation of <u>4a</u> with 5% Rh on alumina gave tricyclic enaminoketones <u>5i</u> and <u>5ii</u>, of which stereoselective elaboration to gephyrotoxin <u>8</u> has already been established.

Diels-Alder cycloaddition of o-quinone methide N-substituted imine intermediate $(2)^{1}$ provides a convenient method for the synthesis of nitrogen-containing polycycles, which, however, has scarcely been exploited, because the general methodology for the generation of 2 is lacking. Recently, the fluoride-anion induced 1,4-elimination, which offered a mild and efficient generation of o-quinodimethanes²⁾, has successfully been extended to o-[N-(trimethyl-silyl)-N-alkylamino]benzyltrimethylammonium halides (1) to generate in situ o-quinone methide N-alkylimines (2)³⁾.



In this paper, we describe a new approach to the synthesis of biologically active alkaloid, gephyrotoxin $(\underline{8})^{4}$ which has been performed on the basis of an intramolecular cycloaddition of the o-quinone methide N-alkenylimine (2a) and the subsequent reductive elaboration of the resulting benzo[e]indolizidine framework.



Preparation of a starting ammonium salt $\frac{1}{10}$ requisite for the generation of o-quinone methide N-alkenylimine (2a) was carried out by N- and O-silylations of $3a^{5}$, followed by quaternization with methyl bromide. Thus, n-BuLi (4.4 mmol, hexane solution) was added to a solution of $3a^{5}$ (2 mmol) in THF (6 mL) and stirred for 6 hr. The stirring solution was treated with trimethylchlorosilane (4.4 mmol) and a catalytic amount of 4-dimethylaminopyridine (25 mg in THF) at 0°C for 3 hr and then at room temperature overnight. The N-trimethylsilylation in the presence of 4-dimethylaminopyridine remarkably improved a yield of the final cyclization product (4) (24% \rightarrow 71%). The reaction mixture was diluted with ether and filtered to remove insoluble materials under nitrogen atmosphere followed by evaporation <u>in vacuo</u>. The residue was dissolved in acetonitrile and stirred with a large excess of methyl bromide (30 mmol) at 0°C to room temperature overnight, and then evaporated <u>in vacuo</u>.

Viscous ammonium bromide la thus prepared was used without purification for the following generation and cyclization of 2a. A solution of the crude la in acetonitrile (10 mL) was added dropwise into a stirring mixture of CsF (6 mmol) and acetonitrile (10 mL) at ca. 65° C over

1.5 hr and then gently refluxed for 2.5 hr. After the reaction mixture was diluted with ether, filtered and evaporated <u>in vacuo</u>, the residue was subjected to preparative thin layer chromatography on silica gel with 1 : 2 acetone-hexane to give benzo[e]indolizidine derivative $\binom{4a}{2}$ (71% based on 3a) [TLC on silica gel R_f=0.45 (1 : 2 acetone-hexane)]. ¹³C NMR spectrum⁶) of 4a produced exhibited two set of signals in a relative ratio of 4~5 : 1, suggesting that 4a consists of a mixture of <u>cis</u> and <u>trans</u> isomers 4a-i and 4a-ii. Attempts to separate the two stereoisomers have failed.

Similarly, cycloaddition of an ammonium bromide <u>lb</u>, in which the hydroxy group has been protected with tetrahydropyran instead of trimethylchlorosilane, afforded a mixture of <u>4a-i</u> and <u>4a-ii</u> in a ratio of 4~5 : 1 (ca. 60% yield) after deprotection.

Next, $\frac{4a}{4a}$ was transformed to a tricyclic enaminoketone $\frac{5}{2}$, which has already been converted to gephyrotoxin ($\frac{6}{20}$) by Kishi^{4a,b)}. Initial attempts to reduce $\frac{4a}{4a}$ by the Birch reaction were not satisfactory. Reduction of $\frac{4a}{4a}$ with metallic lithium in liquid ammonia containing THF and <u>tert</u>amyl alcohol did not give any products, resulting in the recovery of $\frac{4a}{4a}$. Use of n-propylamine instead of ammonia in the reduction produced the desired $\frac{5}{2}$ but only in a low yield (~10%). High-yielding reduction to $\frac{5}{2}$ has been effected by hydrogenation (70 atm., 70°C, 1.5 hr) of $\frac{4b}{7}$ ⁷⁾ using 5% Rh on alumina in anhydrous ethanol to afford $\frac{5}{2}$ (75%) [TLC on silica gel R_f=0.50(4 : 1 acetone-methanol)] together with a small amount of $\frac{7}{2}$ and the hydrogenolysis product $\frac{6}{2}$ ($\frac{6+7}{2} < 10\%$) after chromatography on silica gel with a 4 : 1 acetone-methanol. ¹H NMR and IR spectra of $\frac{5}{2}$ produced were almost consistent with those of $\frac{5-1}{2}$, not $\frac{5-1}{2}$, reported by Kishi^{4a)}. But ¹³C NMR exhibited two set of signals for the product $\frac{5}{2}$ in a ratio of $5 : 1^{80}$, which may be assignable to <u>cis-51</u> and <u>trans-511</u>, respectively. Separation of the two stereoisomers of $\frac{5}{2}$ was difficult. The desired stereoisomer is most conveniently isolated at the stage of $\frac{7}{2}^{4a}$, which is obtained by stereoselective hydrogenation of $\frac{5}{2}$ with 5% Pt on alumina in anhydrous ethyl acetate.

Transformation of the tricyclic enaminoketone 5 to gephyrotoxin 8 via 7 has already been established by Kishi ^{4a,b)}. Consequently, a sequence of reactions mentioned above presents a formal total synthesis of gephyrotoxin.

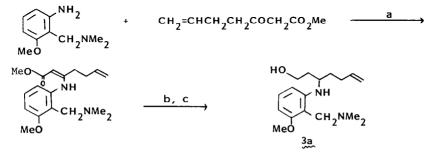
References and Notes

- 1) a) E. M. Burgess and L. McCullagh, J. Am. Chem. Soc., 88, 1580 (1966).
 - b) M. Fisher and F. Wagner, Chem. Ber., 102, 3486 (1969).
 - c) Y.-I. Mao and V. Boekelheide, J. Org. Chem., 45, 1547 (1980).
 - d) M. Lancaster and D. J. H. Smith, J. Chem. Soc., Chem. Commun., 471 (1980).

2) a) Y. Ito, M. Nakatsuka and T. Saegusa, J. Am. Chem. Soc., 104, 7609 (1982).

b) Y. Ito, Y. Amino, M. Nakatsuka and T. Saegusa, J. Am. Chem. Soc., in press.

- 3) Y. Ito, S. Miyata, M. Nakatsuka and T. Saegusa, J. Am. Chem. Soc., 103, 5250 (1981).
- 4) a) R. Fujimoto, Y. Kishi and J.F. Blount, J. Am. Chem. Soc., 102, 7156 (1980).
 - b) R. Fujimoto and Y. Kishi, Tetrahedron Letts, 4197 (1981).
 - c) L.E. Overman and C. Fukaya, J. Am. Chem. Soc., 102, 1454 (1980).
- 5) Compound 3a (bp 130-132°C/0.1 mm Hg) was prepared in 69% overall yield according to the following scheme.



- a: refluxed in benzene containing AcOH b: NaBH₃CN in CH₃OH, pH 4 c: LiAlH₄ in Et₂O
- 6) 4a-i: ¹³C NMR (CDCl₃) δ 19.97, 26.13, 28.91, 30.71, 35.25, 54.45, 55.79, 59.39, 97.15, 103.49, 108.70, 126.05, 143.94, 156.35. 4a-ii: ¹³C NMR (CDCl₃) δ 21.00, 27.48, 28.56, 29.77, 35.03, 53.82, 57.32, 59.71, 97.33, 103.85, 109.15, 144.12, 156.97. 4a: Anal. Calcd for $C_{15}H_{21}NO_{2}$ C, 72.84; H, 8.56; H, 5.66. Found C, 72.55; H, 8.83; N, 5.72.
- 7) Compound 4b [TLC on silica gel $R_f=0.45$ (10:7 acetone-hexane)] was prepared in 75% yield by treating 4a with BBr₃ in methylene chloride at 0° to room temperature for 2 hr.
- 8) 51: ¹³C NMR (CDC1₃) 818.85, 20.69, 25.99, 26.35, 27.74, 29.36, 34.35, 37.01, 55.66, 56.34, 57.50, 103.80, 159.27, 191.50. 511: ¹³C NMR (CDC1₃) 819.08, 20.19, 26.17, 27.02, 28.42, 29.19, 34.76, 37.50, 55.08, 58.08, 58.35, 105.33, 159.45, 192.22.

Acknowledgment. This work was supported by the Ministry of Education, Science and Culture (Grant-in-Aid for Special Project Research No. 57118002). We are grateful to Prof. K. Narasaka of the University of Tokyo for useful discussions and suggestions.

(Received in Japan 7 April 1983)